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MARKSCHEME

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answers and video explanations to these questions here; Remember that on Paper 2 you can use your data book.

IB Chemistry SL Sample Test Paper 2 - Studynova

SL Paper 2 Chloroethene, C_2H_3Cl , is an important organic compound used to manufacture the polymer poly(chloroethene). Draw the Lewis structure for chloroethene and predict the H-C-Cl bond angle.

IB Questionbank - GitHub Pages

SL Paper 2 Molten sodium chloride can be electrolysed using graphite electrodes. Draw the essential components of this electrolytic cell and identify the products that form at each electrode. Product formed at positive electrode (anode):

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Chemistry Standard level Paper 2
Candidate session number 14 pages 1
hour 15 minutes Thursday 12 May 2016

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(morning) 2216 - 6105 Instructions to candidates y Write your session number in the boxes above. y Do not open this examination paper until instructed to do so. y Answer all questions. y Write your answers in the boxes provided.

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CIE Paper 2 IGCSE Chemistry Past Papers - PMT

#1: Complete papers 1 and 2 in one sitting. The IB Chemistry SL and IB Chemistry HL papers are intense, forcing you to sit and concentrate for two hours for SL and 3 hours and 15 minutes for HL. You need to build up your mental strength so you don't make careless errors by the end of paper 2.

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May 2015 Chemistry Standard level Paper 2

Examination (standard level) There are three papers: Paper 1 (multiple choice) - 20%; Paper 2 (structured questions) - 40%; Paper 3 (data and options) - 20%; Total - 80%; Paper 1 (SL): 45 minutes. 30 multiple-choice questions on core, about 15 of which are common with HL. The questions on paper 1 test assessment objectives 1, 2 and 3.

IB Chemistry- Assessment in the Diploma Program

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SL Paper 2. Explain why: Define the term first ionization energy. [2] ... Second mark can be scored even if reagent is incorrect. ... and there is no set-formula in relation to question setting. Organic chemistry is an integral part of the IB SL Chemistry programme, and is covered in Topic 10 of the guide (12 hours in total). Hence, candidates ...

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Predict the direction in which the equilibrium will shift when the following changes occur. The volume increases.

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The temperature decreases. $\text{H}_2\text{O(g)}$ is removed from the system.

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SL Paper 2 Consider the following equilibrium. $2\text{SO}_2(\text{g}) + \text{O}_2(\text{g}) \rightleftharpoons 2\text{SO}_3(\text{g})$
 $\Delta H^\ominus = -198 \text{ kJmol}^{-1}$ Deduce the equilibrium constant expression, K_c , for the reaction.

IB Questionbank

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The one lingering question mark is the
offensive line. This is a big one too. On
paper, once again, this group is more
talented than last year. ... A virtual
offseason doesn't build chemistry on ...

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