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Are your answers to parts a and b consistent with the information in Model 1? yes. STOP POGILTM Activities for High School Chemistry . Model 3 - Number of Protons and Attractive Force Force of Attraction (Newtons) 0.10 nm 2.30 x 10<sup>-8</sup> 0.10 nm X 10<sup>-8</sup> 0.10 nm 6.90 x

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counter-example from Model 1 that supports your answer. 3 (so fijpes w 12. Considering your answers to Question 1 1, write a definition of isotope using a grammatically correct sentence. Your group must come to consensus on this definition. S sax-ne h D-ç bt.đ oc Q rwasS STOP 13.

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HS Chemistry POGIL Activity Page 3 Basic Stoichiometry Model 3 Given the following equation:  $\text{___ N}_2 (\text{g}) + \text{___ H}_2 (\text{g}) \rightarrow \text{___ NH}_3 (\text{g})$  9. If 0.052 mol  $\text{N}_2$  are reacted, how many mol  $\text{NH}_3$  are formed? Using dimensional analysis, show how you calculated your answer. 10.

## **HS Chemistry POGIL Activity Basic Stoichiometry**

Download Free Pogil Oxidation And Reduction Answer Key. oxidation numbers to each species in the equation.  $\text{Cu} + 2\text{N} + 5\text{O}^{-2} + \text{Al} \rightarrow \text{Al} + 3\text{N} + 5\text{O}^{-2} + \text{Cu}^0$ . Step 2: Identify the species with an increase in charge....

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5. The net charge on a sulfide ion ( $S^{2-}$ ) is  $-2$ . Explain how this ion obtains its charge. Your answer should include a discussion of subatomic particles. Answers will vary. There are two extra electrons in the ion compared to a neutral S atom. There are two more electrons than protons in the  $S^{2-}$  ion. Two electrons have been added to an atom of ...

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